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Huan-Mei Guo,^a Fang-Fang Jian,^b* Pu-Su Zhao,^b Yun-Chen Zhang^a and Yu-Feng Li^b

^aDepartment of Chemistry, Weifang College, Weifang 261061, People's Republic of China, and ^bNew Materials and Function Coordination Chemistry Laboratory, Qingdao University of Science and Technology, Qingdao 266042, People's Republic of China

Correspondence e-mail: ffj2003@163169.com

Key indicators

Single-crystal X-ray study T = 293 KMean σ (C–C) = 0.003 Å R factor = 0.041 wR factor = 0.112 Data-to-parameter ratio = 17.0

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5-(2-Chlorophenyl)-3-(4-chlorophenyl)-1-phenyl-2-pyrazoline

In the title structure, $C_{21}H_{16}Cl_2N_2$, the pyrazoline ring makes dihedral angles of 6.76 (9), 7.52 (10) and 81.88 (10)°, respectively, with the 4-chlorophenyl ring, the phenyl ring and the 2-chlorophenyl ring. The 5-phenyl ring is almost perpendicular to the pyrazoline ring.

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Comment

Pyrazoline-based fluorophores have attracted much interest because of their simple structure and favorable photophysical properties, such as large extinction coefficient and quantum yield (Rivett *et al.*, 1983). The published data identify pyrazoline fluorophores as a platform for the development of PET-based cation fluorescent sensors (Fahrni *et al.*, 2003). Among them, 1,3,5-trisubstituted pyrazolines can be easily prepared from phenylhydrazine and chalcone derivatives (Nakamichi *et al.*, 2002). We report the crystal structure of the title compound, (I).



In (I) (Fig. 1), the C=N bond length [1.297 (2) Å] is longer and the N1-N2 bond length [1.380 (2) Å] is shorter than those in the similar structures reported by Rurack *et al.* (2000), Kimura *et al.* (1977) and Ge (2006) [C=N = 1.291 (2), 1.283 (2) and 1.293 (3) Å, respectively, and N-N = 1.394 (3), 1.390 (3) and 1.384 (2) Å, respectively]. The C-Cl bond lengths [1.750 (2) and 1.751 (2) Å] are slightly longer than that in a similar structure (1.745 Å; Kimura *et al.*, 1977). All the bond lengths and angles fall in the normal ranges. The pyrazoline ring makes dihedral angles of 6.76 (9), 7.52 (10) and 81.88 (10)°, respectively, with the *p*-chlorophenyl, phenyl and 2-chlorophenyl rings.

Experimental

© 2007 International Union of Crystallography All rights reserved Compound (I) was synthesized by reaction of phenylhydrazine (0.02 mol) and 1-(4-chlorophenyl)-3-(2-chlorophenyl)-2-propenyl-1-



Figure 1

The molecular structure of (I), showing the atom-labeling scheme, with displacement ellipsoids drawn at the 30% probability level.

ketone (0.02 mol) in acetic acid solution (40 ml). Single crystals of (I) suitable for X-ray measurements were obtained by recrystallization from EtOH at room temperature.

Crystal data

$C_{21}H_{16}Cl_2N_2$	V
$M_r = 367.26$	Z
Triclinic, P1	D
a = 8.3520 (17) Å	Μ
b = 9.5020 (19) Å	μ
c = 11.751 (2) Å	Т
$\alpha = 89.57 \ (3)^{\circ}$	В
$\beta = 80.43 \ (3)^{\circ}$	0.
$\gamma = 81.83 \ (3)^{\circ}$	

Data collection

Enraf-Nonius CAD-4 diffractometer ω scans Absorption correction: none 4139 measured reflections 3865 independent reflections $V = 910.1 (3) Å^{3}$ Z = 2 $D_{x} = 1.340 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 0.36 \text{ mm}^{-1}$ T = 293 (2) KBlock, yellow 0.45 × 0.44 × 0.44 mm

3098 reflections with $I > 2\sigma(I)$ $R_{int} = 0.013$ $\theta_{max} = 27.0^{\circ}$ 3 standard reflections every 100 reflections intensity decay: none Refinement

\mathbf{D} for a set \mathbf{r}	$1/[-2/(E^2)] + (0.0479 \text{ P})^2$
Rennement on F	$w = 1/[\sigma^{-}(F_{o}^{-}) + (0.04/8P)^{-}]$
$R[F^2 > 2\sigma(F^2)] = 0.041$	+ 0.3091P]
$wR(F^2) = 0.112$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.06	$(\Delta/\sigma)_{\rm max} < 0.001$
3865 reflections	$\Delta \rho_{\rm max} = 0.28 \ {\rm e} \ {\rm \AA}^{-3}$
227 parameters	$\Delta \rho_{\rm min} = -0.48 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
	Extinction coefficient: 0.238 (8)

H atoms were positioned geometrically and allowed to ride on their parent atoms, with C-H distances in the range 0.93–0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *NRCVAX* (Gabe *et al.*, 1989); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL/PC* (Sheldrick, 1990); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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